Voluntary Use and Exposure Information Profile Perfluorooctane Sulfonic Acid and Various Salt Forms

CHEMICAL IDENTIFICATION I.

Chemical Name:

Perfluorooctane Sulfonates

CAS Number:

Various, including: 1763-23-1 (acid)

2908 1-56-9 (ammonium salt)

70225-14-8 (DEA salt) 2795-39-3 (K-salt) 29457-72-5 (Li Salt)

The perfluorooctane sulfonate anion (PFOS) has no specific CAS number. The listed salts summarize the designated Tier I perfluorochemicals that are all considered perfluoroctane sulfonates.

II. **COMPANY IDENTIFICATION**

Company Name: 3M

Site Locations:

10746 Innovation Road 1)

Cottage Grove, MN 55016

2) 1400 State Docks Road

Decatur, AL 35601

Technical Contact: W.A. Weppner

Phone:

65 1/733-6374

Address:

3M Center, Building 236-1B-10

St. Paul. MN 55 144

III. **ON-SITE ACTIVITIES**

∉ AS	Mfg. (1997)	<u>Imported</u>
1763-23-1 29081-56-9	Less Than	0
70225-14-8	200,000 lb/yr.	0
2795-39-3" 29457-72-5		0

Estimate the amount of subject chemical distributed off-site:

95% of manufacture/import

^{*}Submitted in 1998 TSCA IUR additional information.

Narrative Description and Process Flow Schematic:

3M Company utilizes a process known as Simons Electra-Chemical Fluorination (ECF) to synthesize organofluorine molecules. In this process, organic feedstocks are dispersed in liquid, anhydrous hydrogen fluoride, and an electric current is passed through the solution, causing the hydrogen atoms on the molecule to be replaced with fluorine. The predominant components of the products created by this process have the same carbon skeletal arrangement as the feedstock used, but with all of the hydrogen atoms replaced by fluorine. However, fragmentation and rearrangement of the carbon skeleton can also occur and significant amounts of cleaved, branched and cyclic structures may be formed. The degree of fluorination of the organic feedstock is also dependent upon the specific carbon chain length of the feedstock and parameters of the ECF process such as electrical current and the length of time the process is run. It is possible to synthesize fully fluorinated or perfluoroorganic molecules where all of the hydrogen atoms of the hydrocarbon feedstock have been replaced by fluorine atoms. Using these perfluoroorganic molecules as basic building blocks, unique chemistries can be created by further reactions with functionalized hydrocarbon molecules.

3M has produced sulfonyl based fluorochemicals commercially for over 40 years using the ECF process. A basic building block of such products and the highest production volume fluorochemical 3M manufactures is perfluoroctanesulfonyl fluoride (POSF). The starting feedstock for this reaction is 1 -octanesulfonyl fluoride. (Reaction 1)

Reaction 1

$$\begin{array}{c} Energy \\ -----> \\ ECF \\ \hline \\ 1\text{-Octanesulfonyl} \quad fluoride \\ \end{array}$$
 Energy
$$------> \\ ECF \\ \hline \\ Perfluorooctanesulfonyl \quad fluoride \\ \end{array}$$

The electrochemical fluorination process yields about 35%-40% straight chain (normal) POSF, and a mixture of biproducts and waste of unknown and variable composition comprised of the following:

- 1) higher and lower straight-chain homologs, i.e., $n-CnF_2n_{+1}SO_2F$, e.g., $C_6F_{13}SO_2F$, $C_7F_{15}SO_2F$, $C_9F_{19}SO_2F$ which comprise about 7% of the process output
- 2) branched-chain, perfluoroalkylsulfonyl fluorides with various chain lengths, about 18-20% of the output
- straight-chain, branched, and cyclic (non-functional) perfluoroalkanes and ethers, which comprise about 20-25% of the output
- 4) "tars" (high molecular weight fluorochemical byproducts) and other byproducts, including molecular hydrogen, which comprise about 10- 15% of the output.

Because of slight differences in process conditions, raw materials, and equipment, the mixture produced by the electrochemical fluorination process varies somewhat from lot-to-lot and from plant-to-plant. The product that results from electrochemical fluorination is thus not a pure chemical but rather a mix of isomers and homologues. The commercialized POSF derived products are a mixture of approximately 70% linear POSF derivatives and 30% branched POSF derived impurities.

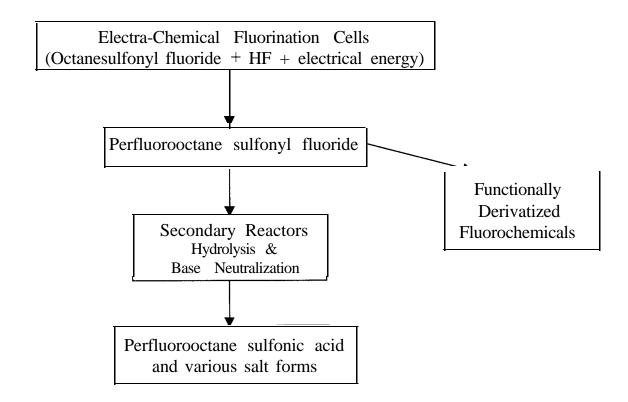
During production, byproducts and waste products are formed. The volatile waste products, such as perfluoromethane, have been vented to the atmosphere in the past, but improvements are underway to capture and destroy these releases by thermal oxidation. The tars are incinerated at an in-house, hazardous waste incinerator. The byproducts, many of which are incompletely fluorinated with hydrogen atoms still present, can be recycled back into the ECF process or are partially degraded in stabilization processes, and discharged to controlled, in-house, wastewater treatment systems. The treatment sludge is either landfilled or land-incorporated, Some of the non-POSF byproducts are recovered and sold for secondary uses.

POSF is itself a commercially viable product, but is primarily an important intermediate in the synthesis of substances used in many other 3M fluorochemical products. The majority is used to produce functionally derivatized fluorochemicals and high molecular weight polymeric products.

The perfluorooctane sulfonic acid manufactured by hydrolyzing POSF intermediates to the sulfonic acid. The various salts of perfluorooctane sulfonate are manufactured by base neutralization of the acid to the appropriate salt in a batch reaction.

The following block flow diagram describes the process discussed above.

BLOCK FLOW DIAGRAM FOR PERFLUOROSULFONIC ACID AND ITS SALTS



IV. SITE RELEASE AND TRANSFER INFORMATION FOR TRI CHEMICALS

Not Applicable Perfluorooctane sulfonic acid and its salts are not listed on TRI.

V. SITE RELEASE AND TRANSFER INFORMATION FOR NON-TRI CHEMICALS

A. ON-SITE AIR RELEASES FROM DECATUR, AL AND COTTAGE GROVE, MN

Estimated Total Estimated % #days/years
Annual Releases Accuracy of release occurs

(lbs. 1997) Estimate (optional)

Fugitive (non-point) Negligible, Not

quantified

Stack (point) 0

Comments:

Physical chemical property information and industrial hygiene monitoring support that PFOS salts would not be in the form of volatile air emissions. Some industrial hygiene testing at Cottage Grove, MN has shown the presence of PFOS in work areas, however the industrial hygiene personnel have indicated that this material is believed to be in the form of dried particulate dust. Although dispersed PFOS has been detected in the work area, the concentrations do not indicate significant contributions to annual site air emissions.

Industrial hygiene data for PFOS salts has not been generated in Decatur, AL because they are in a liquid slurry versus particle form.

Fugitive air emissions testing for the Decatur wastewater treatment system was conducted in late 1999. Results are still under review and no final report is available. However, no PFOS was detected in air samples.

B. WATER RELEASES FROM SITE

Estimated Total Estimated %
Annual Releases Accuracy of
(lbs. 1997) Estimate (optional)

Water releases: 10.000

Number of days/year release occurs: (See comments below)

Receiving Water Name: Baker's Creek at the junction with the Tennessee River

NPDES Number: ALD004023 164

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Comments:

Although manufacturing of the salts listed on this form is associated with less than 10 wastewater discharges per year, the hydrolysis of other sulfonated compounds to PFOS in the wastewater treatment system result in an ongoing discharge of the PFOS compounds. The calculated discharge amount therefore is not attributed to just the salt production. A percentage of the discharges are from the PFOS salt manufacturing. There is not data to distinguish between PFOS discharge from PFOS salt manufacturing versus hydrolysis of other sulfonated compounds manufactured at this site to PFOS.

PFOS in wastewater discharge is not measured as specific salt compounds, but instead reflects the total amount of dissociated salts and compounds that hydrolyze to perfluorooctane sulfonate.

COTTAGE GROVE, MN

Estimated Total Estimated %
Annual Releases (lbs. 1997)

Estimated %
Accuracy of
Estimate (optional)

Water releases: 1000

Number of days/year release occurs: 11 times per year

Receiving Water Name: Mississippi

NPDES Number: MN0000 1449

Comments: The discharge amount is based upon 'per batch estimates of releases'. Removal of the PFOS to the sludge from the Cottage Grove, MN wastewater system was calculated based upon the measured removal of PFOS from wastewater in the clarifiers at Decatur, AL.

C. ON-SITE LAND RELEASES

DECATUR, AL

Estimated Total Estimated %
Annual Releases (lbs. 1997)

Estimated %
Accuracy of
Estimate (optional)

Landfill 0
Land Treatment/Land Amendment 40,000
Surface Impoundments Not quantified Underground Injection 0
Other (specify):

Comments:

In 1997, land-amended sludges were held in an impoundment until weather permitted application to the field. The impoundment is no longer used as part of the normal wastewater treatment operation except during equipment downtime. A percentage of the PFOS land treatment is from the PFOS salt manufacturing. There is not data to distinguish between PFOS land treatment from PFOS salt manufacturing versus hydrolysis of other sulfonated compounds manufactured at this site for PFOS.

The land treatment of Decatur sludge was discontinued in early 1998. Sludge now is transported to an offsite landfills, after passing through a thickener and a sludge press.

Estimated PFOS levels in the sludge are based upon the reduction of PFOS in the wastewater stream as it passes through the clarifiers at the site. The PFOS levels in the sludge have been reduced by changes in the amount of materials being discharged to the sewer. Wastewater data from 1999 show that PFOS levels in the sludge were less than 20,000 pounds per year.

COTTAGE GROVE, MN

	Estimated Total Annual Releases (lbs. 1997)	Estimated % Accuracy of Estimate (optional)
Landfill	0	
Land Treatment/Land Amendmen	nt 0	
Surface Impoundments	0	
Underground Injection	0	
Other (specify):		

Comments: Sludge from the Cottage Grove facility is sent to an off-site industrial landfill.

D. **OFF-SITE TRANSFERS** – Decatur, AL and Cottage Grove, MN (1997)

D1. Transfer to Publicly Owned Treatment Works (POTW)

Number of days/year the release occurs: Not applicable – waste is not sent to a POTW

Annual Transfer (lb): 0

Estimated % Accuracy of Transfer Estimate (optional) (%):

POTW Name:
Street Address:

Not Applicable
Not Applicable

City:NotApplicableCountry:NotApplicableState:NotApplicableZipCode:NotApplicable

NPDES Number: Not Applicable

Comments:

CAS Number 1763-23-1, 29081-56-9, 70225-14-8, 2795-39-3, 29457-72-5 / 3M Company / 4/27/2000 **D2. TRANSFERS TO OTHER OFF-SITE LOCATIONS**

DECATUR, AL

Estimated Total Annual Releases (lbs. 1997) Estimated %
Accuracy of
Estimate (optional)

Incineration: Not applicable

Wastewater Treatment (Excluding POTW)

Not applicable

Underground Injection Not applicable

Hazardous Waste (RCRA Subtitle C) landfill

Not applicable

Other Landfill Not applicable

Recycle or Recovery Not applicable

Unknown or Other

Comments:

COTTAGE GROVE, MN

Estimated Total Annual Releases (lbs. 1997) Estimated %
Accuracy of
Estimate (optional)

Incineration: 0 – See comment

Wastewater Treatment (Excluding POTW)

Not applicable

Underground Injection Not applicable

Hazardous Waste (RCRA Subtitle C) landfill

Not applicable

Other Landfill 7400

Recycle or Recovery Not applicable

Unknown or Other

Comments: Cottage Grove facility utilizes incineration for all their drummed wastes. Waste summary does not show any drummed PFOS salt wastes in 1997.

Sludge from the Cottage Grove facility is sent to an industrial landfill.

VI. ON-SITE WORKPLACE EXPOSURE

CAS Number: <u>2795-39-3</u> Company: <u>3M Company</u>, <u>Specialty Materials Manufacturing</u>; <u>Division</u>, Cottage Grove, MN

(potassium nerfluorooctane sulfonate)

This information will assist EPA in characterizing the number of workers potentially exposed and the magnitude, frequency, and duration of potential exposure. When providing monitoring data, ensure that data is linked with worker activities described in question 2.

1. Estimate the number of workers potentially exposed routinely to the subject chemical for each of the exposure duration times. If a worker is involved in more than one activity, enter only his/her most typical activity in the table. Don't count a worker more than once. The total number in the table should equal the total number of workers potentially exposed.

Hours/Day	Days/Year				
	<10	10-100	100-250	>250	
<.25					
.25-1	1 2	23			
1-8	6	23			
>8					

2. Describe the routine worker activities to which the workers in question 1 are exposed: sampling, removal of filter cake, and drumming of liquids, manufacture an article, etc. For these activities, describe the physical state of the subject chemical (liquid, gas, particulate, or aerosol, etc.) and, if in a mixture, the chemical's concentration:

The chemical (2795-39-3) is a light, free-flowing powder. Initially, the material is a wet slurry (within a reactor) and is then transferred to drying equipment for final processing and packaging. The above employees operate and maintain the drying equipment, collect quality samples, and also package (by gravity or by hand scooping) dry powder from the spray equipment into drums, pails, and cans.

3. Provide industrial hygiene monitoring data, if available, with a brief description of the sampling method and exposure scenario monitored, e.g., describe the specific worker activities performed by the individuals monitored. For privacy considerations, please do not include any personal identifiers such as a worker's name or social security number with any data submitted to EPA.

These samples were collected during operation of drying equipment and packaging activities. Past sample collection and analysis has utilized 37-mm Nucleopore filter cassettes with either gravimetric, inductively coupled plasma emission spectroscopy (for K+), or liquid chromatography analysis. The current method is an OSHA Versatile Sampler (OVS) tube analyzed by liquid chromatography/mass spectrometry. Wipe sample data for the chemical does exist and has indicated significant levels of this material on surfaces in the production area. The results were used for focusing decontamination and exposure control efforts.

Area/Job	Sample	# of	Minimum	Maximum	Geometric	Geometric
Classification	Type	Samples	Concentration	Concentration	Mean	Standard
						Deviation
Production						
Unit:						
Chemical	Partial Shift	10	0.00013	40 mg/m3	0.050 mg/m	64.407
Process	(Sample		mg/m3			
Operator	Time					
	Range: 7-					
	125					
	minutes)					
Pilot Plant:						
Technician	Partial Shift	1	0.45 mg/m3	0.45 mg/m3	NA	NA
	(Sample					
	Time: 70					
	minutes)					

4. Briefly describe the engineering controls used to minimize exposure to this chemical:

The drying and packaging activities are contained within small-room enclosures with negative pressure and filtered exhaust ventilation. Local exhaust may be present directly at the point of packaging.

5. Briefly list the personal protective equipment your workers regularly wear to prevent exposure of this chemical:

Employees working within the above small-room enclosures use airline respirators, chemical-protective coveralls (Tyvek), boot covers, and gloves.

CAS Number: <u>1763-23-1</u> Company: <u>3M Comuany, Snecialty Material Manufacturing Division,</u> Cottage Grove, MN

Chemical Name	CAS No.
Perfluorooctane sulfonic acid	1763-23-1

This information will assist EPA in characterizing the number of workers potentially exposed and the magnitude, frequency, and duration of potential exposure. When providing monitoring data, ensure that data is linked with worker activities described in question 2.

1. Estimate the number of workers potentially exposed routinely to the subject chemical for each of the exposure duration times. If a worker is involved in more than one activity, enter only his/her most typical activity in the table. Don't count a worker more than once. The total number in the table should equal the total number of workers potentially exposed.

Hours/Day	Days/Year					
	<10 10-100 100-250 >250					
<.25						
.25-1		6				
1-8		12				
>8						

2. Describe the routine worker activities to which the workers in question 1 are exposed: sampling, removal of filter cake, and drumming of liquids, manufacture an article, etc. For these activities, describe the physical state of the subject chemical (liquid, gas, particulate, or aerosol, etc.) and, if in a mixture, the chemical's concentration:

The chemical formulation containing 1763-23-1 is a clear colorless liquid solution (32-38% concentration). Employee activities include operation and maintenance of process equipment, product quality sampling and drumming as an intermediate or final product liquid solution containing 1763-23-1.

3. Provide industrial hygiene monitoring data, if available, with a brief describe the specific worker activities performed by the individuals monitored. For privacy considerations, please do not include any personal identifiers such as a worker's name or social security number with any data submitted to EPA.

No data available.

4. Briefly describe the engineering controls used to minimize exposure to this chemical:

Local exhaust ventilation used at product container openings during container filling.

CAS Number 1763-23-l) 2908 1-56-9, 70225-14-8, 2795-39-3, 29457-72-5 / 3M Company / 4/27/2000

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5. Briefly list the personal protective equipment your workers regularly wear to prevent exposure of this chemical:

Employees wear PVC jackets and pants, neoprene gloves, and face shield and chemical splash goggles for containerization of solutions containing 1763-23-l.

CAS Number: <u>70225-14-g</u> Company: <u>3M Company, Specialty Materials Manufacturing Division, Cottage Grove. MN</u>

Chemical Name	CAS No.
Diethanolamine salt of perfluorooctane	70225-14-g
sulfonate	

This information will assist EPA in characterizing the number of workers potentially exposed and the magnitude, frequency, and duration of potential exposure. When providing monitoring data, ensure that data is linked with worker activities described in question 2.

1. Estimate the number of workers potentially exposed routinely to the subject chemical for each of the exposure duration times. If a worker is involved in more than one activity, enter only his/her most typical activity in the table. Don't count a worker more than once. The total number in the table should equal the total number of workers potentially exposed.

Hours/Day	Days/Year					
	<10 10-100 100-250 >25					
<.25						
.25-1	2					
1-8	4					
>8						

2. Describe the routine worker activities to which the workers in question 1 are exposed: sampling, removal of filter cake, and drumming of liquids, manufacture an article, etc. For these activities, describe the physical state of the subject chemical (liquid, gas, particulate, or aerosol, etc.) and, if in a mixture, the chemical's concentration:

The chemical (70225-14-g) is a clear, light colored solution containing 19-23% 70225-14-g. Employee activities include operation and maintenance of process equipment, product quality sampling and drumming of final liquid solution containing 70225-14-g.

Provide industrial hygiene monitoring data, if available, with a brief description of the sampling method and exposure scenario monitored, e.g., describe the specific worker activities performed by the individuals monitored. For privacy considerations, please do not include any personal identifiers such as a worker's name or social security number with any data submitted to EPA.

No data available.

4. Briefly describe the engineering controls used to minimize exposure to this chemical:

Local exhaust ventilation used at product container openings during packaging.

5. Briefly list the personal protective equipment your workers regularly wear to prevent exposure of this chemical:

Employees wear PVC gloves and chemical splash goggles for containerization of product.

CAS Number: <u>29081-56-g</u> Company: <u>3M Company</u>, <u>Snecialty Materials Manufacturing Division</u>, Cottage Grove, MN

Chemical Na	CAS No.		
Ammonium	perfluorooctane	sulfonate	29081-56-9

This information will assist EPA in characterizing the number of workers potentially exposed and the magnitude, frequency, and duration of potential exposure. When providing monitoring data, ensure that data is linked with worker activities described in question 2.

1. Estimate the number of workers potentially exposed routinely to the subject chemical for each of the exposure duration times. If a worker is involved in more than one activity, enter only his/her most typical activity in the table. Don't count a worker more than once. The total number in the table should equal the total number of workers potentially exposed.

Hours/Day	Days/Year				
	<10	10-100	100-250	>250	
<.25					
.25-1	1 2				
1-8	12				
>8					

2. Describe the routine worker activities to which the workers in question 1 are exposed: sampling, removal of filter cake, and drumming of liquids, manufacture an article, etc. For these activities, describe the physical state of the subject chemical (liquid, gas, particulate, or aerosol, etc.) and, if in a mixture, the chemical's concentration:

The chemical formulation containing 2908 1-56-9 is a clear amber solution (19-23% concentration). Employee activities include transferring wet slurry to filter press, operating filter press, transferring filter cake to reactor, and drumming final liquid solution containing 2908 1-56-9. The above employees also operate and maintain process equipment and collect quality samples.

3. Provide industrial hygiene monitoring data, if available, with a brief description of the sampling method and exposure scenario monitored, e.g., describe the specific worker activities performed by the individuals monitored. For privacy considerations, please do not include any personal identifiers such as a worker's name or social security number with any data submitted to EPA.

No data available.

4. Briefly describe the engineering controls used to minimize exposure to this chemical:

Local exhaust available at the filter press, at the reactor opening and at the product container openings.

5. Briefly list the personal protective equipment your workers regularly wear to prevent exposure of this chemical:

Employees wear PVC jackets and pants, neoprene gloves, and chemical splash goggles. Half mask dust-mist respirators are used for handling of wet filter cake.

CAS Number: <u>29457-72-5</u> Company: <u>3M Company, Snecialty Materials Manufacturing Division,</u> Cottage Grove, MN

Chemical Name			CAS No.
Lithium	perfluorooctane	sulfonate	29457-72-5

This information will assist EPA in characterizing the number of workers potentially exposed and the magnitude, frequency, and duration of potential exposure. When providing monitoring data, ensure that data is linked with worker activities described in question 2.

1. Estimate the number of workers potentially exposed routinely to the subject chemical for each of the exposure duration times. If a worker is involved in more than one activity, enter only his/her most typical activity in the table. Don't count a worker more than once. The total number in the table should equal the total number of workers potentially exposed.

Hours/Day	Days/Year						
	<10	· · · · · · · · · · · · · · · · · · ·					
<.25							
.25-1	12						
1-8	6						
>8							

2. Describe the routine worker activities to which the workers in question 1 are exposed: sampling, removal of filter cake, and drumming of liquids, manufacture an article, etc. For these activities, describe the physical state of the subject chemical (liquid, gas, particulate, or aerosol, etc.) and, if in a mixture, the chemical's concentration:

The chemical formulation containing 29457-72-5 is a clear amber solution (24-26% concentration). Employee activities include operation and maintenance of process equipment, product quality sampling and drumming of final liquid solution containing 29457-72-5.

3. Provide industrial hygiene monitoring data, if available, with a brief description of the sampling method and exposure scenario monitored, e.g., describe the specific worker activities performed by the individuals monitored. For privacy considerations, please do not include any personal identifiers such as a worker's name or social security number with any data submitted to EPA.

No data available.

4. Briefly describe the engineering controls used to minimize exposure to this chemical:

Local exhaust ventilation used at product container openings during packaging.

CAS Number 1763-23-1, 29081-56-9, 70225-14-8, 2795-39-3, 29457-72-5 / 3M Company / 4/27/2000

5. Briefly list the personal protective equipment your workers regularly wear to prevent exposure of this chemical:

Employees wear PVC jackets, neoprene gloves, and chemical splash goggles for containerization of product.

CAS Number: <u>2795-39-3</u> Company: <u>3M Company</u>, <u>Suecialty Materials Manufacturing Division</u>, Decatur, AL

Chemical Name		CAS No.	
Potassium	perfluorooctane	sulfonate	2795-39-3

This information will assist EPA in characterizing the number of workers potentially exposed and the magnitude, frequency, and duration of potential exposure. When providing monitoring data, ensure that data is linked with worker activities described in question 2.

1. Estimate the number of workers potentially exposed routinely to the subject chemical for each of the exposure duration times. If a worker is involved in more than one activity, enter only his/her most typical activity in the table. Don't count a worker more than once. The total number in the table should equal the total number of workers potentially exposed.

Hours/Day	Days/Year			
	<10	10-100	100-250	>250
<.25				
.25-1			24	
1-8				
X 3				

Describe the routine worker activities to which the workers in question 1 are exposed: sampling, removal of filter cake, and drumming of liquids, manufacture an article, etc. For these activities, describe the physical state of the subject chemical (liquid, gas, particulate, or aerosol, etc.) and, if in a mixture, the chemical's concentration:

The chemical (2795-39-3) is in an aqueous solution (25%). Employees operate and maintain the production equipment and collect quality samples. Material is produced, stored, and metered into product container in a closed system

3. Provide industrial hygiene monitoring data, if available, with a brief description of the sampling method and exposure scenario monitored, e.g., describe the specific worker activities performed by the individuals monitored. For privacy considerations, please do not include any personal identifiers such as a worker's name or social security number with any data submitted to EPA.

No data available.

4. Briefly describe the engineering controls used to minimize exposure to this chemical:

Enclosed process.

5. Briefly list the personal protective equipment your workers regularly wear to prevent exposure of this chemical:

Employees use safety glasses and neoprene gloves.

CAS Number: <u>1763-23-1</u>, <u>29081-56-9</u>, <u>29457-72-5</u>, <u>70225-14-8</u> Company: <u>3M - Decatur</u>, <u>AL</u>

Chemical Name	CAS No.
Perfluorooctane sulfonic acid	1763-23-1
Ammonium perfluorooctane sulfonate	29081-56-9
Lithium perfluorooctane sulfonate	29457-72-5
Diethanolamine salt of perfluorooctane	70225-14-8
sulfonate	

None of these materials are manufactured at 3M-Decatur.

VII. CHEMICAL END USES

A. END USE AS AN INTERMEDIATE CONSUMED TO MARE OTHER CHEMICALS

Al. On-Site Use as a Intermediate:

CAS #1763-23-1

Perfluorooctane sulfonic acid is used as an intermediate while the salts are not used as intermediates. The perfluorooctane sulfonate salts are only used as finished products.

Product chemical class or product chemical (Include CAS number if appropriate)		% of total* volume of subject chemical manufactured or imported
1.	Fluorochemical trivalent chromium compound	1-5
2.	Perfluorooctane sulfonate salt	1-5
		*As reported in Part III, p.2
A2.	Off-Site Use as an Intermediate:	

Product chemical class or product chemical (Include CAS number if appropriate)

1. Not Applicable

% of total volume of subject chemical manufactured or imported*

Not Applicable

*As reported in Part III, p.2

B. END USES OTHER THAN AS A CONSUMED INTERMEDIATE

The following two tables present a summary of the information contained in VII.B regarding uses of perfluorooctane sulfonic acid and its salt forms.

Table 1 details the end applications in which a particular CAS Number is used. For each CAS Number, the % used in each application totals to 100%.

Table 1. CAS Number by Application

CAS NUMBER	APPLICATION	
2795-39-3	1. Film former in Fire Fighting Foams	
	2. Mist suppressant for acid baths in metal plating factories	
	3. Surfactant in acid etching processes for production of circuit	
	boards	
29457-72-5	1. Surfactant in alkaline cleaners	
	2. Emulsifier in commercial floor polish	
	3. Mist suppressant for acid baths in metal plating factories	
	4. Surfactant in acid etching processes for production of circuit	
	boards	
I	5. Registered active ingredient for insect control. Used in	
	formulations used in bait traps	
70225-14-8	1. Emulsifier in commercial floor polishes	
	2. Mist suppressant for acid baths in metal plating factories	
	3. Surfactant in acid etching processes for production of circuit	
	boards	
29081-56-9	1. Surfactant in acid etching processes for production of circuit	
	boards	
1763-23-1	No commercial products	

Table 2 details the multiple CAS Numbers which may be used in any one application.

Table 2. Application by CAS Number

CAS NUMBER	APPLICATION	
2795-39-3	Film former in Fire Fighting Foams	
2795-39-3	Mist suppressant for acid baths in metal plating factories	
29457-72-5		
70225-14-8		
2795-39-3	Surfactant in acid etching processes for production of circuit	
29457-72-5	boards	
70225-14-8		
29081-56-9		
29457-72-5	Emulsifier in commercial floor polishes	
70225-14-8		
29457-72-5	Surfactant in alkaline cleaners	
29457-72-5	Registered active ingredient for insect control. Used in	
	formulations used in bait traps	
1763-23-1	No commercial products	

CAS Number 2795-39-3, Potassium perfluorooctane sulfonate

Use Number 1 of 3

Description of Chemical End Use: Chemical used as film former which is blended into tire fighting foams. These products are designed for primary use by fire fighting professionals.

Percent of total manufactured or imported Volume going to this use: 92 ± 2

If used in a mixture check appropriate box To indicate weight fraction. Average Values are acceptable:

Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent – wet solid
Gas or vapor
X Liquid solution
Other (Explain)

Use Number 2 of 3

Description of Chemical End Use: Metal plating factories use this chemical to suppress oxidizing acid mist in order to protect their workers' health.

Percent of total manufactured or imported Volume going to this use: $\underline{6} \pm \underline{2}$

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent wet solid
Gas or vapor
Liquid solution
Other (Explain)

Use Number 3 of 3

Description of Chemical End Use: Electronic factories add this chemical as a surfactant to improve the wetting of strong acids used in the etching of precise patterns in a silicon water or a printed circuit board.

Percent of total manufactured or imported Volume going to this use: $\underline{2} \pm \underline{0.5}$

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

<u>X</u> <1% (<1000 ppm) 1-30% 30-60% 60-90% >90% Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent - wet solid
Gas or vapor
X Liquid solution
Other (Explain)

CAS Number 29457-72-5, Lithium perfluorooctane sulfonate

Use Number 1 of 5

Description of Chemical End Use: Chemical formulators use this chemical as a surfactant to improve the wetting of water-based products that are sold as alkaline cleaners. These cleaners are then diluted and sold to the consumer or contract cleaner.

Percent of total manufactured or imported Volume going to this use: 23.5 ± 0.5

Check all physical forms of the chemical during this use:

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent — wet solid
Gas or vapor
X Liquid solution
Other (Explain)

Use Number 2 of 5

Description of Chemical End Use: Metal plating factories use this chemical as a surfactant to suppress oxidizing acid mist in order to protect their workers' health.

Percent of total manufactured or imported Volume* going to this use: $\underline{1} \pm \underline{1}$

If used in a mixture check appropriate box
To indicate Weight fraction. Average
Values are acceptable:

Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent – wet solid
Gas or vapor
X Liquid solution
Other (Explain)

Use Number 3 of 5

Description of Chemical End Use: Electronics factories add this chemical as a surfactant to improve the wetting of strong acids used in the etching of precise patterns in a silicon water or a printed circuit board.

Percent of total manufactured or imported Volume going to this use: $\underline{1} \pm \underline{1}$

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent – wet solid
Gas or vapor
X Liquid solution
Other (Explain)

Use Number 4 of 5

Description of Chemical End Use: Chemical formulators use this chemical as a surfactant to improve the wetting of water-based products that are sold as commercial floor polishes.

Percent of total manufactured or imported Volume going to this use: 23.5 ± 0.5

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent - wet solid
Gas or vapor
X Liquid solution
Other (Explain)



Use Number 5 of 5

Description of Chemical End Use: Chemical formulators use this chemical as a registered active ingredient for insect control. It is blended with bait that is attractive to insects and placed in bait stations. Either commercial facilities or consumers can use these bait stations.

Percent of total manufactured or imported Volume going to this use: 53 ± 0.5

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

- X - <1% 1-30% 30-60% 60-90% >90% Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent-wet solid
Gas or vapor
X Liquid solution
Other (Explain)

CAS Number 29081-56-9, Ammonium perfluorooctane sulfonate

Use Number 1 of 1

Description of Chemical End Use: Electronics factories add this chemical as a surfactant to improve the setting of strong acids used in the etching of precise patterns in a silicon wafer or a printed circuit board.

Percent of total manufactured or imported Volume going to this use: 100

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent – wet solid
Gas or vapor
X_ Liquid solution
Other (Explain)

CAS Number 70225-14-8, Diethanol amine perfluorooctane sulfonate

Use Number 1 of 3

Description of Chemical End Use: Chemical formulators use this chemical as a surfactant to improve the emulsification of materials used in commercial and consumer floor polishes.

Percent of total manufactured or imported Volume going to this use: $\underline{35} \pm \underline{2}$

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent – wet solid
Gas or vapor
Liquid solution
Other (Explain)

Use Number 2 of 3

Description of Chemical End Use: Metal plating factories use this chemical to suppress oxidizing acid mist in order to protect their workers' health.

Percent of total manufactured or imported Volume going to this use: 48.5 ± 3

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent -wet solid
Gas or vapor
Liquid solution
Other (Explain)

Use Number 3 of 3

Description of Chemical End Use: Electronic factories add this chemical as a surfactant to improve the wetting of strong acids used in the etching of precise patterns in a silicon water or a printed circuit board.

Percent of total manufactured or imported Volume going to this use: 16.5 ± 2

If used in a mixture check appropriate box To indicate Weight fraction. Average Values are acceptable:

Check all physical forms of the chemical during this use:

Aerosol
Dry Powder
Pellets or large crystals
Water or solvent wet solid
Gas or vapor
X Liquid solution
Other (Explain)